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E-[4-(β-D-Allopyranosyloxy)phenyl]-1-(4-chlorophenyl)prop-2-enone ethanol solvate

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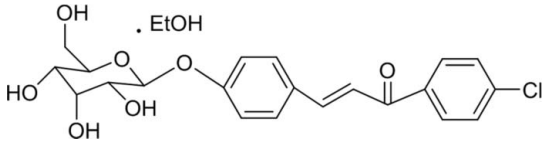
Key indicators: single-crystal X-ray study; T = 292 K; mean σ(C–C) = 0.006 Å; R factor = 0.048; wR factor = 0.143; data-to-parameter ratio = 10.0.

The title compound, $C_{21}H_{21}ClO_7\cdot C_2H_5OH$ was synthesized by the condensation reaction between helicid [systematic name: 4-(β-D-allopyranosyloxy)benzaldehyde] and 4-chloroacetophenone in ethanol. In the molecular structure, the pyranoside ring adopts a chair conformation. In the crystal structure, the molecules are linked by intermolecular O—H···O hydrogen bonds involving the OH groups from the pyranoside unit and from the ethanol solvent molecule.

Related literature

For helicid, see: Chen *et al.* (1981) and for its biological activity, see: Sha & Mao (1987). For the pharmacological activity of some helicid derivatives, see: Fan *et al.* (2007).



Experimental

Crystal data

$C_{21}H_{21}ClO_7\cdot C_2H_6O$
M_r = 466.90
Monoclinic, $P2_1$
a = 11.000 (5) Å
b = 7.712 (3) Å
c = 13.213 (4) Å
β = 92.08 (2)°
V = 1120.2 (8) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.22 mm^{-1}
T = 292 K
0.48 × 0.44 × 0.36 mm

Data collection

Enraf–Nonius CAD-4 diffractometer
Absorption correction: for a sphere [*WinGX*; Farrugia, 1999)]
T_{min} = 0.903, T_{max} = 0.926
2921 measured reflections
2815 independent reflections
2305 reflections with $I > 2\sigma(I)$
R_{int} = 0.008
3 standard reflections
every 200 reflections
intensity decay: 1.4%

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.048
$wR(F^2)$ = 0.143
S = 1.10
2815 reflections
281 parameters
1 restraint
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.43 e Å^{-3}
$\Delta\rho_{min}$ = −0.30 e Å^{-3}
Absolute structure: Flack (1983), 562 Friedel pairs
Flack parameter: 0.14 (12)

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O2—H2*O*···O3^{i}	0.82	2.02	2.809 (4)	160
O4—H4*O*···O2^{ii}	0.82	1.88	2.694 (4)	171
O5—H5*O*···O8^{ii}	0.82	1.97	2.696 (5)	148

Symmetry codes: (i) $-x+1, y-\frac{1}{2}, -z$; (ii) $x, y+1, z$.

Data collection: *DIFRAC* (Gabe & White, 1993); cell refinement: *DIFRAC*; data reduction: *NRCVAX* (Gabe *et al.*, 1989); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

The authors thank Mr Zhi-Hua Mao of Sichuan University for the X-ray data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2216).

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supplementary materials

Acta Cryst. (2009). E**65**, o634 [**doi:10.1107/S1600536809006424**]

E-[4-(β-D-Allopyranosyloxy)phenyl]-1-(4-chlorophenyl)prop-2-enone ethanol solvate

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Comment

The natural compound helicid, 4-(β-D-allopyranosyloxy)benzaldehyde, which is extracted from the fruit of *Helicia nilagirica Beed* (Chen *et al.*, 1981) is a major active ingredient of Chinese herbal medicine. It has good biological effects on the central nervous system with a low toxicity (Sha & Mao, 1987). Some helicid derivatives have been reported with good pharmacological activities (Fan *et al.*, 2007). The title compound, a new helicid derivative, was synthesized *via* reaction of helicid and 4-chloroacetophenone, in good yield.

In the molecule of the title compound (Fig. 1), the average of C—C bond lengths in the six-membered pyranoside ring is 1.522 (5) Å. The average C(sp^3)—O and C(sp^2)—O bond lengths are 1.414 (5) and 1.392 (4) Å, respectively. The pyranoside ring adopts a chair conformation with hydroxyl group at C3 in axial position and the other substituents at C1, C2 and C4, in equatorial positions.

In the crystal packing, the molecules are connected by intermolecular O—H···O hydrogen bonds (Table 1) involving O1 and O7 atoms and the hydroxyl groups in the main molecule and in the ethanol solvent molecule, forming a three-dimensional network.

Experimental

To a solution of helicid (1.420 g, 5 mmol) in ethanol (20 ml), 10% aqueous solution of sodium hydroxide was added until helicid was dissolved completely. Then 4-chloroacetophenone (0.847 g, 5.5 mmol) was added dropwise, with the vessel placed in an ice bath. The reaction was monitored by TLC. After the reaction completed, the mixture was cooled to room temperature, and then neutralized with diluted hydrochloric acid. The solution was extracted three times with ethyl acetate, and the combined organic layers were dried with anhydrous Na_2SO_4, filtered, and evaporated *in vacuo* to get the crude product. The title compound was recrystallized twice from ethanol, and colourless single crystals were finally obtained by slow evaporation of an ethanol solution, at room temperature.

Refinement

H atoms were positioned geometrically (C—H = 0.93–0.98 Å, O—H = 0.82 Å) and refined using a riding model, with U_{iso}(H) = 1.2U_{eq} (methylene C, aromatic C) or U_{iso}(H) = 1.5U_{eq} (O). The absolute configuration was determined by refinement of a Flack parameter, based on 562 measured Friedel pairs (Flack, 1983), and is in agreement with the expected configuration from the synthetic route.

Figures

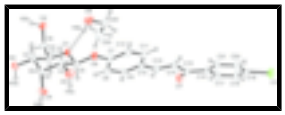


Fig. 1. The molecular structure of (I), with displacement ellipsoids drawn at the 30% probability level.

E-[4-(β-D-Allopyranosyloxy)phenyl]-1-(4-chlorophenyl)prop-2-enone ethanol solvate

Crystal data

$C_{21}H_{21}ClO_7 \cdot C_2H_6O$	$F_{000} = 492$
$M_r = 466.90$	$D_x = 1.384$ Mg m^{-3}
Monoclinic, $P2_1$	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
Hall symbol: P 2yb	Cell parameters from 19 reflections
$a = 11.000$ (5) Å	$\theta = 4.5$–7.4°
$b = 7.712$ (3) Å	$\mu = 0.22$ mm^{-1}
$c = 13.213$ (4) Å	$T = 292$ K
$\beta = 92.08$ (2)°	Block, colourless
$V = 1120.2$ (8) Å^3	0.48 × 0.44 × 0.36 mm
$Z = 2$	

Data collection

Enraf–Nonius CAD-4 diffractometer	$R_{int} = 0.008$
Radiation source: fine-focus sealed tube	$\theta_{max} = 25.5^\circ$
Monochromator: graphite	$\theta_{min} = 1.5^\circ$
$T = 292$ K	$h = -13 \rightarrow 13$
ω/–2θ scans	$k = -9 \rightarrow 9$
Absorption correction: for a sphere [WinGX; Farrugia, 1999)]	$l = -15 \rightarrow 15$
$T_{min} = 0.903$, $T_{max} = 0.926$	3 standard reflections
2921 measured reflections	every 200 reflections
2815 independent reflections	intensity decay: 1.4%
2305 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.048$	$w = 1/[\sigma^2(F_o^2) + (0.099P)^2 + 0.0279P]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.143$	$(\Delta/\sigma)_{max} = 0.001$
$S = 1.10$	$\Delta\rho_{max} = 0.43$ e Å^{-3}
2815 reflections	$\Delta\rho_{min} = -0.30$ e Å^{-3}

281 parameters

1 restraint

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Extinction correction: SHELXL97 (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.056 (7)

Absolute structure: Flack (1983), 562 Friedel pairs

Flack parameter: 0.14 (12)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Cl1	1.19712 (12)	0.6652 (2)	1.29357 (8)	0.0730 (4)
O1	0.6580 (2)	0.7216 (3)	0.27293 (17)	0.0385 (6)
O2	0.5789 (3)	0.4472 (3)	0.1477 (2)	0.0478 (7)
H2O	0.5624	0.4470	0.0867	0.072*
O3	0.4614 (2)	0.8598 (4)	0.05711 (19)	0.0469 (7)
H3O	0.4201	0.9331	0.0848	0.070*
O4	0.5518 (3)	1.1087 (3)	0.1920 (2)	0.0452 (7)
H4O	0.5665	1.2117	0.1830	0.068*
O5	0.8179 (2)	1.1280 (4)	0.2200 (2)	0.0481 (7)
H5O	0.7732	1.2009	0.2443	0.072*
O6	0.8220 (2)	0.8419 (4)	0.35453 (16)	0.0423 (6)
O7	0.7388 (3)	0.5708 (5)	0.9598 (2)	0.0564 (8)
C1	0.5463 (3)	0.7382 (5)	0.2130 (2)	0.0353 (8)
H1	0.4880	0.8055	0.2513	0.042*
C2	0.5718 (4)	0.8348 (5)	0.1143 (3)	0.0376 (8)
H2	0.6262	0.7634	0.0745	0.045*
C3	0.6344 (3)	1.0062 (5)	0.1375 (3)	0.0386 (8)
H3	0.6536	1.0646	0.0742	0.046*
C4	0.7517 (3)	0.9739 (5)	0.2010 (3)	0.0373 (8)
H4	0.8030	0.8947	0.1633	0.045*
C5	0.7157 (3)	0.8832 (5)	0.2974 (3)	0.0379 (8)
H5	0.6621	0.9570	0.3365	0.046*
C6	0.4973 (4)	0.5577 (5)	0.1971 (3)	0.0431 (9)
H6A	0.4793	0.5079	0.2623	0.052*
H6B	0.4217	0.5640	0.1571	0.052*
C7	0.8102 (3)	0.7956 (5)	0.4555 (3)	0.0401 (9)
C8	0.6991 (3)	0.7808 (6)	0.4999 (3)	0.0468 (10)
H8	0.6269	0.8012	0.4631	0.056*
C9	0.6987 (3)	0.7340 (6)	0.6024 (3)	0.0475 (9)
H9	0.6247	0.7247	0.6338	0.057*
C10	0.8041 (3)	0.7015 (5)	0.6579 (3)	0.0423 (9)
C11	0.9136 (4)	0.7145 (6)	0.6094 (3)	0.0497 (10)
H11	0.9861	0.6905	0.6450	0.060*
C12	0.9159 (4)	0.7624 (6)	0.5098 (3)	0.0490 (10)
H12	0.9901	0.7725	0.4787	0.059*
C13	0.7923 (3)	0.6580 (6)	0.7646 (3)	0.0469 (9)
H13	0.7131	0.6442	0.7855	0.056*
C14	0.8794 (3)	0.6351 (6)	0.8365 (3)	0.0454 (9)
H14	0.9610	0.6407	0.8206	0.054*

C15	0.8446 (3)	0.6010 (5)	0.9412 (3)	0.0409 (9)
C16	0.9379 (3)	0.6058 (5)	1.0255 (3)	0.0379 (8)
C17	0.9023 (4)	0.5637 (5)	1.1225 (3)	0.0444 (9)
H17	0.8234	0.5253	1.1321	0.053*
C18	0.9827 (4)	0.5781 (6)	1.2042 (3)	0.0480 (10)
H18	0.9586	0.5490	1.2688	0.058*
C19	1.0986 (4)	0.6360 (5)	1.1897 (3)	0.0453 (9)
C20	1.1381 (4)	0.6725 (6)	1.0948 (3)	0.0489 (10)
H20	1.2180	0.7066	1.0859	0.059*
C21	1.0571 (3)	0.6578 (5)	1.0122 (3)	0.0420 (9)
H21	1.0829	0.6830	0.9475	0.050*
O8	0.7509 (3)	0.3817 (5)	0.3469 (3)	0.0700 (10)*
H8A	0.6815	0.3984	0.3242	0.105*
C22	0.7467 (4)	0.3290 (7)	0.4506 (3)	0.0590 (11)*
H22A	0.7872	0.4154	0.4931	0.071*
H22B	0.7903	0.2204	0.4596	0.071*
C23	0.6220 (6)	0.3068 (9)	0.4826 (5)	0.0845 (16)*
H23A	0.5798	0.2276	0.4377	0.127*
H23B	0.5812	0.4168	0.4810	0.127*
H23C	0.6231	0.2614	0.5503	0.127*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0730 (8)	0.1004 (10)	0.0442 (6)	0.0021 (8)	−0.0177 (5)	−0.0029 (6)
O1	0.0469 (14)	0.0345 (11)	0.0336 (12)	0.0010 (12)	−0.0078 (11)	0.0031 (11)
O2	0.0706 (19)	0.0373 (14)	0.0350 (14)	0.0064 (14)	−0.0040 (14)	−0.0004 (11)
O3	0.0485 (16)	0.0526 (16)	0.0388 (14)	0.0065 (14)	−0.0115 (12)	0.0005 (13)
O4	0.0499 (16)	0.0336 (13)	0.0525 (16)	0.0062 (12)	0.0054 (12)	0.0011 (12)
O5	0.0461 (15)	0.0473 (15)	0.0513 (16)	−0.0075 (13)	0.0087 (12)	−0.0051 (13)
O6	0.0363 (13)	0.0620 (16)	0.0284 (12)	0.0025 (13)	−0.0022 (10)	0.0014 (12)
O7	0.0444 (16)	0.084 (2)	0.0410 (14)	−0.0089 (16)	0.0028 (12)	0.0011 (15)
C1	0.0376 (19)	0.0380 (17)	0.0299 (17)	0.0017 (15)	−0.0021 (15)	0.0000 (15)
C2	0.045 (2)	0.0357 (17)	0.0313 (17)	0.0040 (17)	−0.0020 (16)	0.0036 (16)
C3	0.048 (2)	0.0377 (18)	0.0303 (17)	0.0028 (17)	0.0023 (15)	0.0054 (15)
C4	0.042 (2)	0.0375 (18)	0.0325 (18)	0.0025 (16)	0.0023 (15)	−0.0031 (15)
C5	0.038 (2)	0.0394 (19)	0.0358 (19)	−0.0007 (16)	−0.0045 (16)	−0.0031 (15)
C6	0.047 (2)	0.0405 (19)	0.042 (2)	−0.0029 (18)	−0.0032 (17)	0.0031 (16)
C7	0.0409 (19)	0.047 (2)	0.0323 (18)	0.0004 (18)	−0.0026 (15)	−0.0035 (15)
C8	0.0351 (19)	0.070 (3)	0.0353 (18)	0.005 (2)	−0.0038 (15)	0.0006 (18)
C9	0.0367 (19)	0.070 (3)	0.0360 (19)	0.0027 (19)	0.0038 (16)	0.001 (2)
C10	0.042 (2)	0.054 (2)	0.0309 (17)	0.0017 (19)	−0.0012 (15)	0.0003 (17)
C11	0.040 (2)	0.072 (3)	0.0367 (19)	0.003 (2)	−0.0089 (16)	0.004 (2)
C12	0.040 (2)	0.070 (3)	0.0373 (19)	−0.001 (2)	0.0023 (16)	−0.0004 (19)
C13	0.045 (2)	0.061 (2)	0.0337 (18)	0.001 (2)	−0.0011 (16)	0.0002 (19)
C14	0.042 (2)	0.061 (2)	0.0332 (18)	−0.004 (2)	0.0007 (16)	0.0007 (18)
C15	0.042 (2)	0.046 (2)	0.0349 (18)	−0.0024 (17)	0.0034 (16)	−0.0016 (16)
C16	0.046 (2)	0.0378 (18)	0.0298 (17)	−0.0009 (17)	0.0019 (15)	0.0024 (15)

C17	0.043 (2)	0.051 (2)	0.040 (2)	0.0053 (18)	0.0057 (16)	0.0038 (17)
C18	0.058 (2)	0.056 (2)	0.0296 (18)	0.009 (2)	0.0033 (17)	0.0043 (17)
C19	0.055 (2)	0.047 (2)	0.0336 (18)	0.003 (2)	−0.0049 (16)	0.0001 (17)
C20	0.045 (2)	0.056 (3)	0.045 (2)	−0.006 (2)	0.0004 (17)	0.003 (2)
C21	0.046 (2)	0.048 (2)	0.0321 (17)	−0.0023 (19)	0.0013 (15)	0.0031 (17)

Geometric parameters (Å, °)

Cl1—C19	1.732 (4)	C9—C10	1.372 (5)
O1—C5	1.431 (4)	C9—H9	0.9300
O1—C1	1.443 (4)	C10—C11	1.389 (5)
O2—C6	1.414 (5)	C10—C13	1.460 (5)
O2—H2O	0.8200	C11—C12	1.368 (6)
O3—C2	1.420 (4)	C11—H11	0.9300
O3—H3O	0.8200	C12—H12	0.9300
O4—C3	1.420 (5)	C13—C14	1.336 (5)
O4—H4O	0.8200	C13—H13	0.9300
O5—C4	1.412 (5)	C14—C15	1.473 (5)
O5—H5O	0.8200	C14—H14	0.9300
O6—C7	1.392 (4)	C15—C16	1.488 (5)
O6—C5	1.404 (4)	C16—C21	1.389 (5)
O7—C15	1.221 (5)	C16—C17	1.392 (5)
C1—C6	1.505 (5)	C17—C18	1.376 (5)
C1—C2	1.537 (5)	C17—H17	0.9300
C1—H1	0.9800	C18—C19	1.371 (6)
C2—C3	1.516 (5)	C18—H18	0.9300
C2—H2	0.9800	C19—C20	1.371 (6)
C3—C4	1.533 (5)	C20—C21	1.388 (5)
C3—H3	0.9800	C20—H20	0.9300
C4—C5	1.519 (5)	C21—H21	0.9300
C4—H4	0.9800	O8—C22	1.431 (5)
C5—H5	0.9800	O8—H8A	0.8200
C6—H6A	0.9700	C22—C23	1.460 (7)
C6—H6B	0.9700	C22—H22A	0.9700
C7—C12	1.368 (5)	C22—H22B	0.9700
C7—C8	1.379 (5)	C23—H23A	0.9600
C8—C9	1.402 (5)	C23—H23B	0.9600
C8—H8	0.9300	C23—H23C	0.9600
C5—O1—C1	114.1 (3)	C8—C9—H9	119.0
C6—O2—H2O	109.5	C9—C10—C11	118.1 (3)
C2—O3—H3O	109.5	C9—C10—C13	117.1 (3)
C3—O4—H4O	109.5	C11—C10—C13	124.8 (3)
C4—O5—H5O	109.5	C12—C11—C10	120.7 (3)
C7—O6—C5	117.9 (3)	C12—C11—H11	119.7
O1—C1—C6	106.7 (3)	C10—C11—H11	119.7
O1—C1—C2	109.3 (3)	C11—C12—C7	120.7 (4)
C6—C1—C2	113.9 (3)	C11—C12—H12	119.7
O1—C1—H1	108.9	C7—C12—H12	119.7
C6—C1—H1	108.9	C14—C13—C10	129.1 (4)

C2—C1—H1	108.9	C14—C13—H13	115.5
O3—C2—C3	111.2 (3)	C10—C13—H13	115.5
O3—C2—C1	109.8 (3)	C13—C14—C15	119.1 (3)
C3—C2—C1	110.3 (3)	C13—C14—H14	120.4
O3—C2—H2	108.5	C15—C14—H14	120.4
C3—C2—H2	108.5	O7—C15—C14	120.4 (3)
C1—C2—H2	108.5	O7—C15—C16	119.4 (3)
O4—C3—C2	107.1 (3)	C14—C15—C16	120.2 (3)
O4—C3—C4	110.7 (3)	C21—C16—C17	118.9 (3)
C2—C3—C4	109.6 (3)	C21—C16—C15	122.8 (3)
O4—C3—H3	109.8	C17—C16—C15	118.3 (3)
C2—C3—H3	109.8	C18—C17—C16	120.7 (4)
C4—C3—H3	109.8	C18—C17—H17	119.7
O5—C4—C5	112.7 (3)	C16—C17—H17	119.7
O5—C4—C3	112.3 (3)	C19—C18—C17	119.4 (4)
C5—C4—C3	107.1 (3)	C19—C18—H18	120.3
O5—C4—H4	108.2	C17—C18—H18	120.3
C5—C4—H4	108.2	C18—C19—C20	121.5 (4)
C3—C4—H4	108.2	C18—C19—Cl1	119.3 (3)
O6—C5—O1	106.1 (3)	C20—C19—Cl1	119.2 (3)
O6—C5—C4	108.6 (3)	C19—C20—C21	119.2 (4)
O1—C5—C4	109.8 (3)	C19—C20—H20	120.4
O6—C5—H5	110.8	C21—C20—H20	120.4
O1—C5—H5	110.8	C20—C21—C16	120.3 (4)
C4—C5—H5	110.8	C20—C21—H21	119.8
O2—C6—C1	113.1 (3)	C16—C21—H21	119.8
O2—C6—H6A	109.0	C22—O8—H8A	109.5
C1—C6—H6A	109.0	O8—C22—C23	112.0 (4)
O2—C6—H6B	109.0	O8—C22—H22A	109.2
C1—C6—H6B	109.0	C23—C22—H22A	109.2
H6A—C6—H6B	107.8	O8—C22—H22B	109.2
C12—C7—C8	120.7 (3)	C23—C22—H22B	109.2
C12—C7—O6	116.4 (3)	H22A—C22—H22B	107.9
C8—C7—O6	122.9 (3)	C22—C23—H23A	109.5
C7—C8—C9	117.8 (3)	C22—C23—H23B	109.5
C7—C8—H8	121.1	H23A—C23—H23B	109.5
C9—C8—H8	121.1	C22—C23—H23C	109.5
C10—C9—C8	122.0 (4)	H23A—C23—H23C	109.5
C10—C9—H9	119.0	H23B—C23—H23C	109.5
C5—O1—C1—C6	−178.6 (3)	C7—C8—C9—C10	−0.7 (7)
C5—O1—C1—C2	57.8 (4)	C8—C9—C10—C11	−0.6 (7)
O1—C1—C2—O3	−177.0 (3)	C8—C9—C10—C13	178.6 (4)
C6—C1—C2—O3	63.7 (4)	C9—C10—C11—C12	1.5 (7)
O1—C1—C2—C3	−54.1 (4)	C13—C10—C11—C12	−177.6 (4)
C6—C1—C2—C3	−173.4 (3)	C10—C11—C12—C7	−1.1 (7)
O3—C2—C3—O4	58.7 (4)	C8—C7—C12—C11	−0.2 (7)
C1—C2—C3—O4	−63.3 (4)	O6—C7—C12—C11	−179.5 (4)
O3—C2—C3—C4	178.9 (3)	C9—C10—C13—C14	−173.8 (4)
C1—C2—C3—C4	56.8 (4)	C11—C10—C13—C14	5.2 (7)

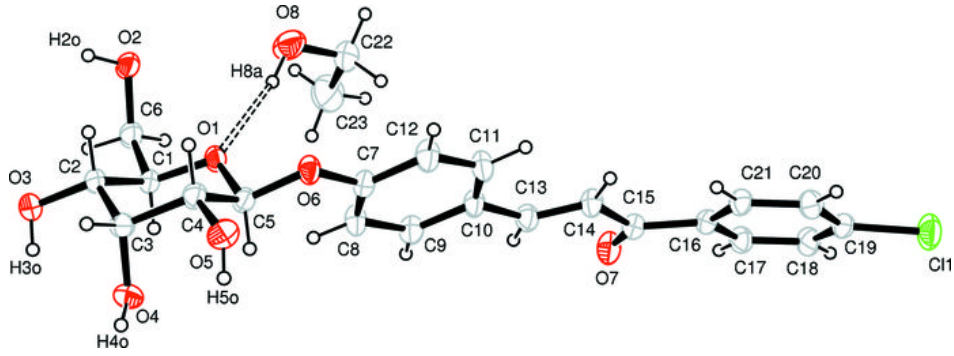
O4—C3—C4—O5	−65.5 (4)	C10—C13—C14—C15	176.6 (4)
C2—C3—C4—O5	176.5 (3)	C13—C14—C15—O7	9.8 (6)
O4—C3—C4—C5	58.7 (4)	C13—C14—C15—C16	−169.3 (4)
C2—C3—C4—C5	−59.2 (4)	O7—C15—C16—C21	−172.7 (4)
C7—O6—C5—O1	−76.9 (4)	C14—C15—C16—C21	6.4 (6)
C7—O6—C5—C4	165.2 (3)	O7—C15—C16—C17	4.3 (5)
C1—O1—C5—O6	−179.6 (3)	C14—C15—C16—C17	−176.6 (4)
C1—O1—C5—C4	−62.5 (4)	C21—C16—C17—C18	1.9 (6)
O5—C4—C5—O6	−59.6 (4)	C15—C16—C17—C18	−175.2 (3)
C3—C4—C5—O6	176.3 (3)	C16—C17—C18—C19	0.5 (6)
O5—C4—C5—O1	−175.2 (3)	C17—C18—C19—C20	−2.9 (6)
C3—C4—C5—O1	60.8 (3)	C17—C18—C19—Cl1	176.6 (3)
O1—C1—C6—O2	−58.8 (4)	C18—C19—C20—C21	2.9 (6)
C2—C1—C6—O2	61.8 (4)	Cl1—C19—C20—C21	−176.6 (3)
C5—O6—C7—C12	−177.7 (3)	C19—C20—C21—C16	−0.4 (6)
C5—O6—C7—C8	3.1 (6)	C17—C16—C21—C20	−2.0 (6)
C12—C7—C8—C9	1.1 (6)	C15—C16—C21—C20	175.1 (4)
O6—C7—C8—C9	−179.7 (4)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O2—H2O···O3^{i}	0.82	2.02	2.809 (4)	160
O4—H4O···O2^{ii}	0.82	1.88	2.694 (4)	171
O5—H5O···O8^{ii}	0.82	1.97	2.696 (5)	148
O3—H3O···O7^{iii}	0.82	2.11	2.741 (4)	134
O3—H3O···O4	0.82	2.41	2.779 (4)	109
O8—H8A···O1	0.82	2.59	2.966 (5)	109

Symmetry codes: (i) −*x*+1, *y*−1/2, −*z*; (ii) *x*, *y*+1, *z*; (iii) −*x*+1, *y*+1/2, −*z*+1.

Fig. 1


O8
C22
H2o
O2
H8a
C6
O1
C23
C12
C11
C1
C7
C2
O3
O6
C13
C21
C20
C15
C4
C5
C10
C14
C16
C3
C8
C9
C19
O5
O7
C17
C18
Cl1
H3o
H5o
O4
H4o